

# Non-Uniform Triple Helical Structure in Chick Skin Type I Collagen on Thermal Denaturation: Raman Spectroscopic Study\*

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The individual chains in the triple helix of collagen occur in a conformation related to polyproline II because of the presence of large number of imino peptide bonds. However, these residues are not evenly distributed in the collagen molecule which also contains many non-imino residues. These non-imino regions of collagen may be expected to show preference for other than triple helical conformations. The appearance of several Raman bands in solution phase at 65 °C raises the possibility of non-uniform triple helical structure in collagen. Raman spectroscopic studies on collagen in the solid state and in solution at a temperature greater than its denaturation temperature, reported here suggest that denatured collagen may exhibit an ensemble of conformational states with yet unknown implications to the biochemical interactions of this important protein component of connective tissues.

## Introduction

Collagen, the most abundant protein in vertebrates, exists in a unique triple helical conformation, in which each of the three intertwined chains can be considered to be a polymer of glycine containing triplets, (X-Y-Gly) (Bhatnagar and Rapaka, 1976 and references cited therein). Collagen contains more imino residues than most other proteins, with proline occurring at position X and hydroxyproline at position Y. The imino residues account for approximately one fourth of all residues and the observed polyproline-like conformation of each chain is ascribed to the restricted,  $\varphi$  rotation about the N-C $\alpha$  bond of the peptide unit involving each imino residue, as well as to the interactions of the imino residues as derived from preliminary

NMR and molecular mechanics studies of (Pro-Pro-Gly) 10 (Bhatnagar *et al.*, 1988). An examination of the primary structure of collagen reveals that the imino residues Pro and Hyp are not evenly distributed along the polypeptide, and extended segments show sparse distribution of these residues. The imino-deficient segments may be expected to have conformational preferences different from the polyproline structure of imino-rich segments. While in the native state of collagen, the predominant stereochemical interactions of the imino residues can be expected to direct the overall conformation, a collapse of the triple helix on denaturation can be expected to relax these constraints, facilitating the local acquisition of other possible conformations. Raman spectroscopy is well suited for monitoring triple helix non-triple helix transition and can be expected to provide insight into structural changes occurring in the triple helical collagen structure.

Previous vibrational spectroscopic studies of imino-deficient and imino-rich model peptides have established characteristic vibrational features of conformationally sensitive amide I and III modes of triple helical structures (Renugopalakrishnan *et al.*, 1984; Diem *et al.*, 1984). More recently FT-

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IR and FT-IR photoacoustic studies have been extended to chick skin type I collagen (Renugopalakrishnan *et al.*, 1989). A low frequency Raman vibrational mode in the region  $300\text{--}400\text{ cm}^{-1}$  was found to be characteristic of triple helical conformation (Renugopalakrishnan *et al.*, 1985). An inelastic neutron-scattering study of type I calf skin collagen (Berney *et al.*, 1987) revealed a number of low frequency vibrational modes at  $296^\circ\text{K}$  and at  $110^\circ\text{K}$  and were interpreted as a consequence of significant structural changes in collagen. The unique triple helical structure in collagen assumes a polyproline II type of conformation and should be expected to impose stereochemical constraints by the occurrence of Pro or Hyp occurring at least once at position X and Y in the repeating triplet sequence, Gly-X-Y, on the amide vibrations of collagen. Raman studies of collagen have been previously reported by Frushour and Koenig (1975) and Goheen *et al.* (1978). The appearance of two amide I vibrational modes was interpreted as arising from two non-equivalent species of C=O groups, "associated with the polar and non-polar regions of the collagen polypeptide chains". A distinction between polar and non-polar regions based on amide I stretching frequency observed in collagen Raman spectrum is hard to justify. In the past, there has been considerable difficulty in identifying amide I vibrations from collagen Raman spectra due to extensive hydration of samples and the virtual impossibility of complete subtraction of water background, strong fluorescence background, especially in the previously reported studies in the literature utilizing earlier versions of Raman spectrometers.

Proteins can be regarded as delicately free energy balanced systems and hence the equilibrium that determines the unique conformation of a protein is the one that exists between the native low energy conformation and the lack of obvious long range order in conformational states present in the denatured state (Anfinsen, 1973). It is this equilibrium that is influenced by thermal denaturation favoring the "random" correlate which is essentially devoid of structural regularity (Bhatnagar and Rapaka, 1976) characteristic of the native low energy conformation. Thermal stability of proteins is a subject of contemporary research in our laboratories and progress has been slow in understanding the free energy change ( $\Delta F$ ) that occur at the

characteristic melting temperature,  $T_m$ , of proteins. Thermal stabilization of proteins by increasing their  $T_m$  is one of our long term goals in order to increase their utility in technological applications of proteins (V. Renugopalakrishnan, U.S. Patent to be submitted) and hence the present study is a part of ongoing research on the thermal denaturation and free energy ( $\Delta F$ ) changes of proteins (Oobtake and Renugopalakrishnan, to be submitted).

Information concerning various conformations that may be generated by segments of collagen after their release from the triple helix is necessary for understanding the many biologically important interactions of collagen. Several of the reactive sites of collagen have been identified as lying entirely on individual chains (Hay, 1984). The ability of certain regions of the molecule to assume distinct conformations is likely to be an important mechanism in the regulation of the specificity of the interactions of collagen with cell surface receptors and with other macromolecules such as fibronectin.

## Material and Methods

Chick skin type I collagen was generously provided by Dr. Jerome Gross, Harvard Medical School and Massachusetts General Hospital, Boston, MA. The experimental protocol for its isolation and characterization have been described elsewhere (Heighberger *et al.*, 1978).

**Solid phase Raman spectra:** To obtain the Raman spectrum of the solid collagen, a small amount ( $\sim 1\text{ mg}$ ) of the material was packed into the indented end of a small metal rod. The exposed surface of collagen was smoothed with a knife blade. The metal rod was then positioned in an assemblage which held the exposed solid in the path of the laser beam in such a way that the specular reflection would not be collected. The scattered light was then frequency-analyzed in order to observe the Raman bands. This configuration was somewhat troublesome in that, even at large Raman shift values,  $>600\text{ cm}^{-1}$ , a large background signal was present due to scattered light from the excitation source. However, this method was found to be superior to that of placing the solid in a capillary tube since true glass-free spectra could be recorded.

Collagen in aqueous solution is denatured at 41 °C; however, collagen fibers when in alignment exhibit a rather broad phase transition characterized by shrinkage at elevated temperatures. The center of this transition is at approximately 55 °C (Ramachandran, 1967). In order to maintain uniformity in these studies, we have selected a temperature of 65 °C for the examination of the denatured state of collagen both in solution and in the solid phase. The elevated temperature was maintained by a hot air blower directed at the sample holder and was regulated by a variable resistance device. All of the collagen spectra presented in the study reported here, solid and solution, were obtained with a Spectra-Physics Argon ion laser (Model 164) operating at 547.9 nm with output power typically about 200mW. A Spex Ramalog Model 1401 spectrophotometer was used. The spectral band width (slit width) was generally about 8  $\text{cm}^{-1}$ . Points in the spectra were taken every 3  $\text{cm}^{-1}$  with counts averaged at each point for ten seconds. Typically 100 scans were performed and the results presented in this paper represent the average of 100 scans. The spectrophotometer was automated by a Digital PDP computer which was used to store, manipulate, and display the data (Town *et al.*, 1981). All of the displayed spectra have been subjected to a standard three-point smoothing procedure. It was also necessary to digitally multiply certain spectra in order to compensate for any artificial intensity variations between sample runs.

**Solution phase spectra:** The solutions used for the Raman study were prepared by dissolving the solid collagen in 0.2 M acetic acid at a concentration level of 14.3 mg/ml. A small amount of this solution was then sealed in a standard melting point capillary which was supported in a Harney-Miller cell whose temperature could be controlled. To obtain the elevated temperature spectra, nitrogen gas was passed first over a heating element onto the capillary tube, then over a chromel-alumel thermocouple junction. Voltage applied to the heating element was automatically controlled by a proportional heating source which monitored the thermocouple output. Even though not essential to this study, the temperature of the sample could be accurately maintained to within a fraction of a degree with this configuration. In order to digitally subtract the solvent contribution from the solution

spectra, spectra of acetic acid were recorded at both 25 °C and 65 °C within the Harney-Miller cell. The solution spectra displayed are those that resulted from the subtraction. All of the other experimental parameters described for the solid phase spectra apply for solution phase spectra.

## Results and Discussion

Raman spectra of collagen in the solid state at 25 °C and 65 °C are shown in Fig. 1 and the frequencies of bands with their assignments are listed in Table I. Solid type I collagen at 25 °C exhibits two bands with roughly equivalent intensities in the amide I region, at 1670  $\text{cm}^{-1}$  and 1690  $\text{cm}^{-1}$ , the 1670  $\text{cm}^{-1}$  band occurs in a frequency range normally ascribed to  $\beta$ -sheet structures (Chirgadze and Nevskaya, 1976; Bandekar and Krimm, 1979 and references cited therein). The previous vibrational spectroscopic studies focused on (Pro-Pro-Gly) (Diem *et al.*, 1984) and on a tripeptide, Ala-Gly-Gly (Renugopalakrishnan *et al.*, 1984), known to pack into a triple helical conformation in the solid state from x-ray crystallographic studies (Subramanian and Lalitha, 1983). The vibrational spectral data from the above synthetic collagen-like polypeptide and the two Raman spectroscopic studies of collagen reported in the literature (Frus-hour and Koenig, 1975; Goheen *et al.*, 1978) have shed some light on the characteristic amide I vi-

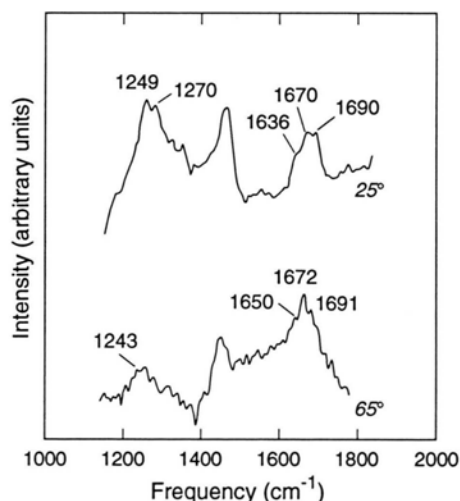


Fig. 1. Raman spectra of type I chicken skin collagen in the solid phase at 25 °C (top) and 65 °C bottom., see text for experimental conditions.

Table I. Major Raman Spectral frequencies ( $\text{cm}^{-1}$ ) of chick skin type I collagen at 25 °C and 65 °C, respectively, in solid and solution phases.

Solid		Solution <sup>b</sup>		Assignment
25 °C	65 °C	25 °C	65 °C	
1249(s)	1243	1257	multiplet structure	<b>amide III</b>
1270(m)	1261	1284(sh)		
1318(sh)				CH 3 symmetric deformation and CH2 Wagging
1346(sh)				
1420(sh)	1460	1434(sh)	1455(sh)	CH 3 assym. deformation and CH2 deformation
1457(s)	1474	1461(s)	1473(a)	
		1473(sh)		
		1527(sh)	1536	<b>amide II</b>
		1542	1559	
		1563(sh)		
1636(sh)	1650(m)	1641(s)	1641(sh)	<b>amide I</b>
1670(s)	1672(s)	1662(m)	1664(s)	
1690(s)	1691(m)	1677(s)		
1695(m)				

<sup>a</sup> Frequencies are accurate to  $\pm 3 \text{ cm}^{-1}$

<sup>b</sup> In acetic acid solution.

s, strong, m, medium, sh, shoulder.

brational modes of collagen. The  $1680 \text{ cm}^{-1}$  band also occurs in solid bovine achilles tendon collagen (Frushour and Koenig, 1975). The barely discernible shoulder at  $1636 \text{ cm}^{-1}$  observed in the present study is low frequency shifted compared to the shoulder at  $1646 \text{ cm}^{-1}$  in solid bovine achilles tendon collagen (Frushour and Koenig, 1975). In contrast, Poly-L-proline II exhibits a strong Raman band at  $1650 \text{ cm}^{-1}$  (Smith, 1969) which has no counterpart in the Raman spectrum of collagen. The band at  $1241 \text{ cm}^{-1}$  with a shoulder at  $1261 \text{ cm}^{-1}$  in the Raman spectrum of poly-L-proline II bears resemblance to the Raman doublet at  $1249 \text{ cm}^{-1}$  (strong) and  $1270 \text{ cm}^{-1}$  (medium) of collagen. The amide III frequencies observed in collagen are indicative of a Ramachandran angle,  $\psi > 90^\circ$  (Lord, 1977). Ramachandran angle,  $\psi > 90^\circ$ , are characteristic of collagen-like structures (Ramachandran and Sasisekharan, 1968). The amide III region of Raman spectra of type I collagen manifests a doublet at  $1249 \text{ cm}^{-1}$  (strong) and  $1270 \text{ cm}^{-1}$  (medium), which are remarkably close to the amide III frequencies at  $1248 \text{ cm}^{-1}$  and  $1271 \text{ cm}^{-1}$  (shoulder) observed in calf skin collagen (Frushour and Koenig, 1975). Raman spectrum of collagen in solid phase at 65 °C, above its  $T_m$  of

41 °C (Privalov *et al.*, 1979; Privalov, 1982), is shown in Fig. 1. On thermal denaturation, the amide I region exhibits a marked difference with the  $1672 \text{ cm}^{-1}$  band gaining intensity which is slightly different from the intense  $1668 \text{ cm}^{-1}$  band and its shoulder at  $1636 \text{ cm}^{-1}$  observed earlier in calf skin collagen (Frushour and Koenig, 1975). The most striking difference on thermal denaturation of collagen can be seen in the amide III region, a conformationally sensitive region of Raman spectra of polypeptides and proteins. The Raman doublet at  $1249 \text{ cm}^{-1}$  and  $1270 \text{ cm}^{-1}$  observed at 25 °C collapse to produce a strong band at  $1243 \text{ cm}^{-1}$  at 65 °C. Raman spectra of collagen in acetic acid solution at 25 °C and 65 °C are shown in Fig. 2 and their frequencies are listed in Table I. In contrast to the observation in the solid phase, collagen solution shows a strong doublet amide I pattern with bands at  $1641 \text{ cm}^{-1}$  and  $1677 \text{ cm}^{-1}$ , respectively. A medium intensity band also occurs at  $1662 \text{ cm}^{-1}$ . The amide III region in contrast has an intense band at  $1257 \text{ cm}^{-1}$  with a shoulder at  $1284 \text{ cm}^{-1}$ . Frushour and Koenig, 1975 have reported only amide III bands for calf skin collagen at a concentration level of 2% in acetic acid solution at 25 °C.

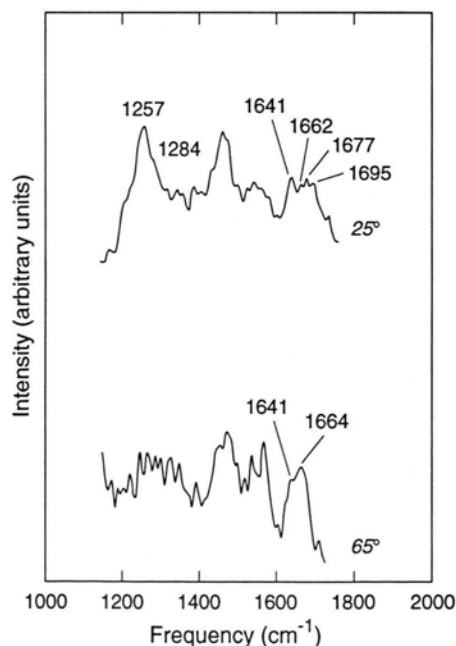


Fig. 2. Raman spectra of type I chicken skin collagen (14.30 mg) in 1 ml. of 0.2 M acetic acid at 25 °C (top) and 65 °C (bottom), see text for experimental conditions.

The amide II region in Raman spectrum of collagen contains a band at  $1542\text{ cm}^{-1}$  with two shoulders at  $1527\text{ cm}^{-1}$  and  $1563\text{ cm}^{-1}$ , respectively. The amide III band on thermal denaturation splits into a multiplet structure exhibiting a complex pattern. The amide II band gains intensity with a doublet at  $1536\text{ cm}^{-1}$  and  $1559\text{ cm}^{-1}$ , respectively. Therefore, it is interesting to observe that Raman studies suggest that the denatured solid collagen manifests an amide I band at  $1672\text{ cm}^{-1}$  and an amide III band at  $1243\text{ cm}^{-1}$  which are archtypical of  $\beta$ -sheet structures (Chirgadze and Nevskaya, 1976 a,b). The complex amide III pattern observed on thermal denaturation of collagen is indicative of several conformational states accessible in the denatured state. The accessibility of conformational states could be due to the loss of ordered structure by the rupture of inter-chain hydrogen bonds, extensive dehydration which will remove water molecules usually associated with collagen structure (see Lim and Griko, 1981; Renugopalakrishnan *et al.*, 1989), and possibly induction of cis-trans isomerization of peptide bonds. Although several conformational states are indicated on the basis of

Raman studies, the denatured collagen may lack long range order. The appearance of several Raman bands in solution phase at  $65^\circ\text{C}$  also raises the possibility of non-uniform triple helical structure unlike  $\alpha$ -helical structures which give rise to a much narrower range of Raman amide I bands,  $1650\text{--}5\text{ cm}^{-1}$  (Renugopalakrishnan and Bhatnagar, 1984). The conclusions derived here may have important implications to the biochemical interactions of this vital protein component of connective tissues.

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